## CARBON DIOXIDE CLUSTERS: (CO2)6 TO (CO2)13

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We recently reported<sup>*a*</sup> assignments of specific infrared bands in the CO<sub>2</sub>  $\nu_3$  region (~2350 cm<sup>-1</sup>) to (CO<sub>2</sub>)<sub>6</sub>, (CO<sub>2</sub>)<sub>7</sub>, (CO<sub>2</sub>)<sub>9</sub>, (CO<sub>2</sub>)<sub>10</sub>, (CO<sub>2</sub>)<sub>11</sub>, (CO<sub>2</sub>)<sub>12</sub>, and (CO<sub>2</sub>)<sub>13</sub>. Spectra are obtained by direct absorption using a rapid-scan tuneable diode laser spectrometer to probe a pulsed supersonic slit-jet expansion and assignments are facilitated by recent calculations of Takeuchi based on the Murthy potential.<sup>*b*</sup> (CO<sub>2</sub>)<sub>6</sub> is a symmetric top with S<sub>6</sub> point group symmetry which can be thought of as a stack of two planar cyclic trimers. (CO<sub>2</sub>)<sub>13</sub> is also an S<sub>6</sub> symmetric top, and consists of a single CO<sub>2</sub> monomer surrounded by an slightly distorted icosahedral cage. The remaining clusters are asymmetric tops without symmetry.

Here we report additional CO<sub>2</sub> cluster results. Calculations based on the SAPT-s potential<sup>c</sup> indicate that the structure of  $(CO_2)_{10}$  may be slightly different from that given by Takeuchi/Murthy. An additional band is observed for each of  $(CO_2)_{13}$  and  $(CO_2)_{10}$ . A feature observed at 2378.2 cm<sup>-1</sup> is assigned as a  $(CO_2)_6$  parallel combination band involving the sum of a fundamental and a low-lying intermolecular vibration. Most significantly, two bands are assigned to a second isomer of  $(CO_2)_6$ . This is also a symmetric top, but now with  $S_4$  symmetry. The two symmetric hexamer isomers observed spectroscopically correspond well with the lowest energy structures given by both the SAPT-s and Murthy intermolecular potentials.

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<sup>&</sup>lt;sup>b</sup>H. Takeuchi, J. Phys. Chem. A 107, 5703 (2008); C.S. Murthy, S.F. O'Shea, and I.R. McDonald, Mol. Phys. 50, 531 (1983).

<sup>&</sup>lt;sup>c</sup>R. Bukowski, J. Sadlej, B. Jeziorski, P. Jankowski, K. Szalewicz, S.A. Kucharski, H.L. Williams, and B.M. Rice, J. Chem. Phys. 110, 3785 (1999).