

AN FTIR STUDY OF THE INFLUENCE OF ENVIRONMENT CONDITIONS ON THE SUCCESSIVE HYDROGENATIONS OF CO

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In a preliminary study, the reaction of CO successive hydrogenation has been performed at 3 and 10 K using a co-injection technique. A H/H₂ mixture and CO molecules are codeposited on a cold mirror and hydrogen atoms are reacting step by step with CO as follows :
CO → HCO → H₂CO → H₃CO → CH₃OH

At 3 K, CO hydrogenation is limited to the first step of the reaction. However, when the reaction is performed at 10 K, CO hydrogenation is total and all intermediates are observed in the IR spectra.

The aim of this present work is to understand the influence of the environment in which the reaction takes place, in order to determine the best conditions promoting CO hydrogenation. Indeed, the catalytic potential of water molecules on CO hydrogenation has been more specifically studied. In this experiment, mixtures of (CO,H₂O) and H/H₂ are co-injected at various concentrations and in a temperature range between 3 and 20 K. It is clear that water environments induce noticeable changes by increasing the chemical reaction rate and promoting HCO conversion into H₂CO.