## ROTATIONAL CONFORMERS OF GROUP VI (Cr, Mo, and W) METAL BIS(TOLUENE) SANDWICH COM-PLEXES

<u>JUNG SUP LEE</u>, SUDESH KUMARI, and DONG-SHENG YANG, *Department of Chemistry, University of Kentucky, Lexington, KY* 40506-0055.

Transition metal bis(arene) sandwich complexes may adopt eclipsed or staggered conformations due to the aromatic ring rotations about the metal-arene axis.<sup>*a*,*b*</sup> In this study, the group VI (Cr, Mo, and W) metal bis(toluene) complexes are synthesized in a laserablation molecular beam source, and their rotational conformers are identified by pulsed-field-ionization zero-electron-kinetic-energy (ZEKE) spectroscopy. For Cr-bis(toluene), the ZEKE spectrum shows three distinctive vibrationless (0-0) transitions between the ground electronic states of the neutral and ionic complexes at 42739(5), 42745(5), and 42805(5) cm<sup>-1</sup>, corresponding to ionization energies of 180°, 60°/120°, and 0° rotamers. In addition, the spectrum exhibits metal-toluene bending (164, 180, 196, and 223 cm<sup>-1</sup>) and stretching (278 and 291 cm<sup>-1</sup>) frequencies of these rotamers. The ground electronic states of the 0° and 180° rotamers are <sup>1</sup>A<sub>1</sub> (C<sub>2v</sub>) and <sup>1</sup>A<sub>g</sub> (C<sub>2h</sub>) in the neutral form and <sup>2</sup>A<sub>1</sub> (C<sub>2v</sub>) and <sup>2</sup>A<sub>g</sub> (C<sub>2h</sub>) in the ionized form, respectively. For the 60° and 120° rotamers, the ground states of the neutral molecules are <sup>1</sup>A (C<sub>2</sub>), and those of the corresponding ions are <sup>2</sup>A (C<sub>2</sub>). Through the variation of the molecular beam conditions, the eclipsed conformer (0°) is determined to be more stable than the staggered ones (180°, 120°, and 60°). Similarly, multiple conformers are identified for the Mo and W complexes.

<sup>*a*</sup>B.S. Sohnlein, S. Li, and D.S. Yang, J. Chem. Phys. 123, 214306 (2005); B.S. Sohnlein and D.S. Yang, J. Chem. Phys. 124, 134305 (2006) <sup>*b*</sup>S.Y. Ketkov, H.L. Sezle, and F.G.N. Cloke, Angew. Chem. Int. Ed. 46, 7072 (2007) and references therein.