THE ROTATIONAL SPECTRUM OF CIS- AND TRANS-HSSOH

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In this talk we present the first pure rotational lines of Oxatrisulfane, HSSOH. Oxatrisulfane is one link between the well-known skew chain molecules HSSSH and HOOOH. The structure of trisulfane HSSSH has been determined using rotational spectroscopy by Liedke *et al.*^{*a*} showing HSSSH to form two stable conformers, *trans* and *cis*, with the *trans*-conformer to be slightly more stable. Only recently hydrogen-trioxide HOOOH has been discovered by Suma *et al.*^{*b*} observing the rotational spectra of the *trans* conformer. For this molecule no evidence for the *cis* structure has been found.

First experimental evidence for the formation of HSSOH was given by Königshofen *et al.*^c via flash vacuum pyrolysis of t-BuS(O)St-Bu at T > 700°C using IR-spectroscopy to monitor the pyrolysis products. We employed this method to synthesize HSSOH and performed first high-resolution millimeterwave measurements in the range of 75–120 GHz by introducing an all solid state spectrometer. More than 170 lines of the *cis* conformer and more than 30 lines of *trans*-HSSOH have been undoubtly identified. Both conformeres are close to the limiting case of a prolate symmetric top with $\kappa_{cis} = -0.925$ and $\kappa_{trans} = -0.924$ respectively. Their permanent dipole moment points mainly along the *b*- and *c*- principle axes and therefore the spectrum displays a clear perpendicular structure. For analyzing the identified lines a standard *S*-reduced Hamiltonian has been used and we were able to fit the rotational parameters *A*, *B*, *C* as well as the centrifugal distortion parameters D_J , D_{JK} , D_K , d_1 and d_2 . The *c*-type transitions of *cis*-HSSOH are approx. 14 times more intense than the corresponding *b*-type transitions, whereas in case of *trans*-HSSOH the intensities of the *b*- and *c*-type transitions are almost equal. All experimentally derived parameters are in excellent agreement with those obtained by high level ab initio calculations^d.

^aLiedtke et al., J. Mol. Struct. 413 265–270 (1997)

^bSuma et al., JACS **127** 14998–14999 (2005)

^cKönigshofen et al., Z. Anorg. Allg. Chem. 625 1779–1786 (1999)

^dJ. Gauss, private communication