

SPECTROSCOPY OF LARGE, COLD MOLECULAR CLUSTERS

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A review will be presented of our work in the spectroscopy of large, cold molecular clusters. Growth of clusters in He droplets at $T = 0.38$ K has been studied.^a Infrared spectra indicate that ammonia clusters consisting of about 10^4 molecules have a compact structure and that inner molecules in the clusters have similar hydrogen-bonded coordination as in crystalline ammonia.^b Compact structures were also obtained in the case of CH_4 and HCl clusters.^c These findings are consistent with ballistic aggregation of particles in the superfluid He droplets. Vibrational and rotational Raman spectra were used to study the state of hydrogen clusters.^d Clusters formed in expansion of a neat para- H_2 gas are solid as evidenced by the vibrational frequency and characteristic splitting of the rotational $S_0(0)$ line. However, clusters of about 10^5 molecules at estimated $T < 1$ K, obtained upon expansion of highly diluted para- H_2 in He ($< 1\%$) have a singular $S_0(0)$ line characteristic of a fluid state. These results offer prospects for observation of superfluidity in hydrogen, which has long been predicted theoretically, but still eludes experimental confirmation.

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