

CHARACTERIZATION OF WEAKLY BOUND $\text{HCF}_3\text{-OCS}$, $\text{H}_2\text{CF}_2\text{-OCS}$, $\text{H}_3\text{CF-OCS}$, AND $\text{HCF}_3\text{-CO}_2$ VAN DER WAALS COMPLEXES BY *AB INITIO* CALCULATIONS AND MICROWAVE SPECTROSCOPY.

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High resolution Fourier-transform microwave spectroscopy has been utilized in a systematic study of the structural and dynamic properties of a series of fluorinated methane molecules complexed to carbonyl sulfide and carbon dioxide. Theoretical data from *ab initio* calculations at the MP2/6-311++G(2d, 2p) level have provided good agreement with the observed experimental values. The structures of $\text{HCF}_3\text{-OCS}$ and $\text{H}_2\text{CF}_2\text{-OCS}$ have been determined from experimentally obtained rotational spectra fit to a Watson *A* reduction Hamiltonian to within 4 kHz. The *a*-type rotational spectrum of $\text{H}_3\text{CF-OCS}$ has been assigned, however, *b*-type transitions still need to be located. The $\text{HCF}_3\text{-CO}_2$ spectrum showed a doubling of transitions into *A* and *E* states (by up to 1.5 MHz), and is currently fit to approximately 20 kHz using XIAM^a. The *ab initio* structures of all the complexes within the series are very similar with $\text{O}\cdots\text{H}$ distances consistently between 2.57-2.65 Å and $\text{O}=\text{C}\cdots\text{C}$ bond angles of 61-64°, although with a wider range of $\text{H}-\text{C}\cdots\text{C}$ angles (*ca.* 60° to 75°). Results for all members of the series will be presented and compared to theoretical data.

^aXIAM v2.5e. H. Hartwig and H. Dreizler, *Z. Naturforsch* **51a**, (1996), 923.