

HIGH RESOLUTION INFRARED SPECTROSCOPY OF THE FLUOROMETHYL RADICAL

E.S. WHITNEY, F. DONG, D.J. NESBITT, *JILA and National Institute of Standards and Technology, University of Colorado, Boulder, CO 80309.*

The CH₂F radical is an intermediate in the decomposition of various atmospheric species and is also important in free radical addition reactions. Previous work on the CH₂F radical has utilized microwave spectroscopy to determine rotational transitions within the ground vibrational state (the symmetric C-H stretch)^a. Infrared spectroscopic methods have also been used to probe the v3 band (C-F stretch) origin and rotational constants^b.

We have used high resolution (0.0005 cm⁻¹) infrared spectroscopy to investigate rovibrational transitions and assign band origins for both the symmetric and asymmetric stretch regions of the CH₂F radical. Measurements were taken using direct absorption (sensitivity = 1 x 10⁻⁷ per root Hz) under slit-jet cooled conditions (20 K). Assignment of the spectrum (up to J=7) has yielded refined lower and upper state rotational and centrifugal distortion constants for this near-prolate top. The ratio of absorption intensities for the symmetric and asymmetric bands provides an interesting comparison to that of the CH₂Cl radical, where no asymmetric band was detected. Interpretation of hyperfine structure will also be discussed.

^aY. Endo, C. Yamada, S. Saito, and E. Hirota, *J. Chem. Phys.* 79(4), 1983.

^bC. Yamada and E. Hirota, *J. Molec. Spec.* 116, 1986.