

EXPERIMENTAL DETERMINATION OF GROUND STATE BINDING ENERGIES IN HETERO- AND HOMONUCLEAR RARE GAS-DIHALOGEN COMPLEXES

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A simple Arrhenius model based on the assumptions of a thermodynamic equilibrium between the T-shaped and linear isomers of the $\text{He} \cdots \text{I}^{35}\text{Cl}(X, v''=0)$ rare gas-heteronuclear dihalogen complex and of a Boltzmann rotational distribution for each isomer is used to find the relative binding energies of the isomers. The model was tested by comparing the intensities of the T-shaped and linear bands observed in laser-induced fluorescence spectra recorded in the $\text{ICl } B-X, 3-0$ spectral region in varying temperature regions in the expansion. The results reveal that the T-shaped isomer is $3.2(1.0) \text{ cm}^{-1}$ higher in energy than the linear isomer. Using two-laser, pump-probe and high-resolution action spectroscopy the linear binding energy was precisely determined to be $21.97(12) \text{ cm}^{-1}$, thereby indicating a T-shaped $\text{He} \cdots \text{I}^{35}\text{Cl}(X, v''=0)$ binding energy of $18.8(1.0) \text{ cm}^{-1}$. The approach was then extended to the $\text{He} \cdots \text{Br}_2(X, v''=0)$ rare gas-homonuclear dihalogen complexes. The intensities of the features associated with transitions of the T-shaped and linear isomers observed in the ro-vibronic spectra recorded in the $\text{Br}_2 } B-X, 11-0$ region are used to monitor changes in the relative populations of the isomers. The T-shaped $\text{He} \cdots \text{Br}_2$ isomer is found to be only $0.7(2) \text{ cm}^{-1}$ higher in energy than the linear isomer. Action spectra reveal that the ground state linear isomer is bound by $17.6(4) \text{ cm}^{-1}$, and thus a binding energy of $16.9(4) \text{ cm}^{-1}$ for the T-shaped isomer is determined. The experimental energies agree remarkably well with those obtained using a two-dimensional $\text{He} + \text{Br}_2(X, v''=0)$ *ab initio* potential energy surface, 17.4 and 16.6 cm^{-1} for the linear and T-shaped isomers, respectively.