

PFI-ZEKE SPECTROSCOPY OF METAL-POLYCYCLIC AROMATIC HYDROCARBONS: Sc-PYRENE and Sc-NAPHTHALENE

JASON F. FULLER, BRAD R. SOHNLEIN, and DONG-SHENG YANG, *Department of Chemistry, University of Kentucky, Lexington, KY 40506-0055.*

Gas-phase complexes of scandium with pyrene ($C_{16}H_{20}$) and naphthalene ($C_{10}H_8$) are produced by using laser ablation and a polycyclic aromatic hydrocarbon (PAH) oven in a pulsed molecular beam source. The resulting Sc-PAH complexes are characterized with pulsed field ionization-zero electron kinetic energy (PFI-ZEKE) photoelectron spectroscopy. The PFI-ZEKE spectrum of Sc-Pyrene exhibits vibrational progressions of 361 and 154 cm^{-1} and a vibrational spacing of 144 cm^{-1} , with the band origin at 40937 cm^{-1} . Through comparison with *ab initio* and Franck-Condon factor calculations, it is determined that this spectral profile arises from a Sc-pyrene complex in which the scandium atom is bound to the “outer” ring in an η^4 - configuration. The 361 cm^{-1} progression is assigned to the Sc^+ -ring stretch and 154 and 144 cm^{-1} features are assigned to pyrene “flying” modes in the cation and neutral, respectively. For Sc-naphthalene, the spectrum shows four vibrational intervals of 389, 347, 298 and 112 cm^{-1} , with the adiabatic ionization energy of 41192 cm^{-1} . A preliminary analysis indicated the Sc atom is in an η^2 bonding mode in this complex. However, a detailed analysis is still in progress.