

THE LOWEST  $^3(n, \pi^*)$  STATE OF 2-CYCLOPENTEN-1-ONE: CAVITY RINGDOWN ABSORPTION SPECTRUM AND RING-BENDING POTENTIAL ENERGY FUNCTION

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The room-temperature cavity ringdown absorption spectra of 2-cyclopenten-1-one (2CP) and deuterated derivatives were recorded near 385 nm. The very weak ( $\epsilon < 1 M^{-1} \text{ cm}^{-1}$ ) band system in this region is due to the  $T \leftarrow S_0$  electronic transition, where  $T$  is the lowest-energy  $^3(n, \pi^*)$  state. The origin band was observed at  $25,963.6 \text{ cm}^{-1}$  for the undeuterated molecule and at  $25,959.4$  and  $25,956.2 \text{ cm}^{-1}$ , respectively, for 2CP- $d_1$  and 2CP- $5,5-d_2$ . For the  $-d_0$  isotopomer, about 50 vibronic transitions have been assigned in a region from  $-500$  to  $+500 \text{ cm}^{-1}$  relative to the origin band. Nearly every corresponding assignment was made in the  $-d_2$  spectrum. Several excited-state fundamentals have been determined for the  $d_0/d_2$  isotopomers, including ring-twisting ( $\nu'_{29} = 238.9/227.8 \text{ cm}^{-1}$ ), out-of-plane carbonyl deformation ( $\nu'_{28} = 431.8/420.3 \text{ cm}^{-1}$ ), and in-plane carbonyl deformation ( $\nu'_{19} = 346.3/330.2 \text{ cm}^{-1}$ ). The ring-bending ( $\nu'_{30}$ ) levels for the  $T$  state were determined to be at  $36.5, 118.9, 213.7, 324.5$ , and  $446.4 \text{ cm}^{-1}$  for the undeuterated molecule. These drop to  $29.7, 101.9, 184.8, 280.5$ , and  $385.6 \text{ cm}^{-1}$  for the  $-d_2$  molecule. A potential energy function of the form  $V = ax^4 + bx^2$  was fit to the ring-bending levels for each isotopic species. The fitting procedure utilized a kinetic energy expansion that was calculated based on the structure obtained for the  $T$  state from *ab initio* calculations. The barrier to planarity, determined from the best-fitting potential energy functions for the  $-d_0, -d_1$ , and  $-d_2$  species, ranges from  $42.0$  to  $43.5 \text{ cm}^{-1}$ . In the  $T$  state, electron repulsion resulting from the spin flip favors nonplanarity. The  $S_0$  and  $S_1$  states have planar structures that are stabilized by conjugation.