

RESONANCE ENHANCED TWO-PHOTON IONIZATION (RE2PI) SPECTRUM OF THE 520 NM SYSTEM OF RbCs

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The rovibrational spectrum of RbCs molecule in the range of 520 nm is observed by the resonance enhanced two-photon ionization (RE2PI) method. A very cold pulsed molecular beam that contains RbCs, Rb₂, Cs₂, etc is generated by a high temperature pulsed nozzle. Only RbCs⁺ ion could be detected using the time-of-flight (TOF) mass spectrometer (mass resolution ~ 1300). Vibrational bands ($v' = 6 \sim 31$) are rotationally resolved by the high resolution dye laser with an intracavity etalon (laser resolution $\sim 0.02 \text{ cm}^{-1}$). The excited electronic state is assigned to the 5 $^1\Sigma^+$ state that dissociates into Rb(5s $^2S_{1/2}$) + Cs(7s $^2S_{1/2}$). By the pseudopotential calculation, the 5 $^1\Sigma^+$ state has adiabatic potential curve with a shelf as a result of the avoided crossing with ionic pair state at long internuclear distances. Observed in the Franck-Condon region, however, the ΔG_v curve shows slightly positive curvature for $v' > 20$, which may result from the avoided crossing with the 6 $^1\Sigma^+$ state at $\sim 6 \text{ \AA}$.

By the selection rules of $\Delta J = \pm 1$ for ${}^1\Sigma \rightarrow {}^1\Sigma$ transitions, only *P* and *R* lines are observed. The rotational constants, B_v , and the vibrational term value, T_v , are determined from the analysis of the rotationally resolved spectra. From the vibrational energy level spacing, ΔG_v , the spectroscopic parameters for the 5 $^1\Sigma^+$ state are determined as $T_e = 18560.12(8) \text{ cm}^{-1}$, $\omega_e = 40.83(1) \text{ cm}^{-1}$, and $\omega_e x_e = 0.2465(6) \text{ cm}^{-1}$.

The dissociation energy, D_e , is 3811.5 cm^{-1} . Measured isotope shifts confirm the absolute vibrational numbering. The B_e and α_e obtained from the B_v vs. v plot are $0.013468(8) \text{ cm}^{-1}$ and $7.48(4) \times 10^{-5} \text{ cm}^{-1}$, respectively. Using these molecular constants, potential energy curve of the 5 $^1\Sigma^+$ state is constructed by the RKR method.