

ROTATIONAL SPECTRA, STRUCTURES AND MODELING OF THE SO₂-OCS, SO₂-CS₂ AND SO₂-CO₂ COMPLEXES

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The rotational spectra of the SO₂-OCS and SO₂-CS₂ weakly bound dimers have been assigned by pulsed-nozzle Fourier-transform microwave spectroscopy. Both complexes have C_S symmetry with the SO₂ straddling the linear monomer. The SO₂ axis makes an angle with the linear monomer of 18.3° and 9.8° respectively in the two complexes. This is in contrast to the structure of SO₂-CO₂.^a This complex has effective C_{2v} symmetry with the SO₂ axis perpendicular to the CO₂. The structural features and dipole moments of these three species will be discussed and compared with results obtained from a semi-empirical model employing electrostatic, dispersion and repulsion interactions.

^aL.H. Sun, I.I. Ioannou and R.L. Kuczkowski, *Mol. Phys.*, **88**, (1996), 255