

THE STRUCTURE, THE INTERNAL ROTATION OF N₂ AND THE ROTATION OF THE ELECTRONIC TRANSITION MOMENT IN THE PARA-DIFLUOROBENZENE-N₂ VAN DER WAALS COMPLEX

QUAN JU, CHARLES S. PARMENTER, *Department of Chemistry, Indiana University, Bloomington IN 47405.*

Cold jet S₁ ← S₀ rotational band contours have been obtained at 0.3 cm⁻¹ resolution for the p-difluorobenzene-N₂ (pDFB-N₂) van der Waals complex and analyzed with an asymmetric top simulation program. Satisfactory simulation is possible only if it is assumed that the in-plane electronic transition moment of pDFB-N₂ rotates about 37° towards the F-F axis from its position normal to that axis in free pDFB. The vdW bond distances from the ring plane to the N₂ bond parallel to that plane are 3.47 Å and 3.42 Å in the S₁ and S₀ states, respectively. Several transitions involving the internal rotation of N₂ about an axis perpendicular to the ring plane have been assigned. The band positions are consistent with two-fold rotational barriers of 14.0 cm⁻¹ and 19.0 cm⁻¹ for the S₀ and S₁ states, respectively. These barriers are in agreement with those recently^a determined from the assignments of different internal rotor transitions seen in the R2PI spectra.

^aY. Hu, W. Lu and S. Yang, *J. Photoch. Photobio. A*, 106, 91(1997).